#### [Tetrahedron 67 \(2011\) 1873](http://dx.doi.org/10.1016/j.tet.2011.01.023)-[1884](http://dx.doi.org/10.1016/j.tet.2011.01.023)

Contents lists available at ScienceDirect

# Tetrahedron

journal homepage: [www.elsevier.com/locate/tet](http://www.elsevier.com/locate/tet)

# Designer peptide dendrimers using click reaction

V. Haridas \*, Yogesh K. Sharma, Srikanta Sahu, Ram P. Verma, Sandhya Sadanandan, Bharat G. Kacheshwar

Department of Chemistry, Indian Institute of Technology, Hauz Khas, New Delhi-110016, India

## article info

Article history: Received 7 November 2010 Received in revised form 19 December 2010 Accepted 8 January 2011 Available online 13 January 2011

Keywords: Peptide Dendrimer Click reaction Triazole Carbohydrate

## ABSTRACT

We designed and synthesized various peptide dendrimers using a 1,3-dipolar cycloaddition (Click) reaction. The dendritic structures reported here include symmetrical, asymmetrical, and cationic dendrimers with triazole, cystine, aromatic, aliphatic, and Lys-Asp dipeptide cores. The high chemoselectivity of the click reaction allowed us to synthesize good yields of high-purity protected and unprotected dendritic structures. Triazole is an excellent peptide bond mimic, which remains hydrolytically stable. Dendrimer 15a and the core unit 21 gelate in a mixture of organic solvents. We also demonstrated the versatility of the design by synthesizing various carbohydrate-based dendrimers. 2011 Elsevier Ltd. All rights reserved.

#### 1. Introduction

Dendrimers are structurally well-defined molecules with a branched architecture and a large number of functional groups at the surface. They also display multivalent interactions (the cluster effect).<sup>1</sup> Dendrimers have applications in diverse areas including nanobiotechnology and tissue engineering.<sup>[2](#page-11-0)-[6](#page-11-0)</sup> The potential applications of dendrimers have generated intense research interest, although there are challenges involved in their synthesis. All synthetic strategies for dendrimers rely on iterative sequences of reaction steps, which yield geometric increases in the number of surface functionalities. Modular reactions are essential for the chemoselective ligation of dendrons required to synthesize fullsized dendrimers. In order to be effectively used for dendrimer synthesis, these modular reactions must proceed efficiently in sterically hindered environments. Recently, 1,3-dipolar cycloaddition reactions have roused considerable interest among researchers because of their nearly quantitative yield, mild reaction conditions, and broad tolerance towards functional groups.

The thermal 1,3-dipolar cycloaddition reaction between alkyne and azide has been known for over a century and was thoroughly investigated by Rolf Huisgen and Koch in the  $1950s<sup>7</sup>$  $1950s<sup>7</sup>$  $1950s<sup>7</sup>$  The reaction is thermodynamically favored ( $\Delta H \sim 55$  kcal/mol) but has a high kinetic energy barrier ( $\sim$  26 kcal/mol). The kinetic stability of alkyne and azide make the reaction very slow at room temperature.

However, the addition of a Copper (I) catalyst accelerates the reaction rate up to  $10<sup>7</sup>$  times and exclusively yields the 1,4 regioisomer (Fig. 1). $8$  These transformations are called 'click' reactions. $9$ 

Tetrahedror

The triazole moiety forms a rigid linking unit that mimics the electronic properties of a peptide bond. The distance between  $R<sup>3</sup>$ and  $R^4$  ([Fig. 2\)](#page-1-0) is 3.9 Å in a peptide bond, but in a triazole structure the distance is approximately 5.0 Å. However, a triazole unit has a higher dipole moment than a peptide, which ensures correct peptide bond mimicry.[10](#page-11-0)

The chemoselective ligation of two unprotected peptide fragments to a full-sized protein is a challenging endeavor. Researchers have developed a number of chemical reactions that can be used to overcome this challenge. The native chemical ligation, $^{11}$  Staudinger, $^{12}$ and thiaproline<sup>[13](#page-11-0)</sup> reactions are elegant and efficient methods of



Fig. 1. 1,3-Dipolar cycloaddition reaction between azide and alkyne.



<sup>\*</sup> Corresponding author. Tel.: þ91 011 26591380; e-mail address: [h\\_haridas@](mailto:h_haridas@hotmail.com) [hotmail.com](mailto:h_haridas@hotmail.com) (V. Haridas).

<sup>0040-4020/\$ -</sup> see front matter  $\odot$  2011 Elsevier Ltd. All rights reserved. doi[:10.1016/j.tet.2011.01.023](http://dx.doi.org/10.1016/j.tet.2011.01.023)

<span id="page-1-0"></span>

Fig. 2. Comparison between the structure of a peptide bond and a triazole unit.

synthesizing proteins. Development of the expressed protein ligation approach further testifies to the combined efforts of chemists and biologists to solve this problem[.14](#page-11-0)

Dendrimers are devoid of secondary structural elements but have compact globular topology. The synthesis of such macromolecular structures, combined with de novo design strategies, may provide protein-like molecules with catalytic and/or other functional properties. However, robust and high-yielding chemical reactions are needed to synthesize protein-like macromolecules effectively. The click reaction is an attractive option for chemo-selectively linking two dendrons.<sup>15–[17](#page-11-0)</sup> Recently, we have used the click reaction to synthesize a variety of peptide dendrimers.<sup>[18](#page-11-0)</sup>

There are many examples of dendrimer synthesis dealing with peptide and non-peptide architecture, but very few synthetic strategies are suitable for generating dendrimers with unprotected functionalities.[19](#page-11-0) In addition to its high yield and high degree of functional group tolerance, the click reaction is advantageous in that it proceeds without altering other functional groups. We envisioned that this reaction could be used to link various protected and unprotected peptide fragments. Triazole is an excellent peptide bond mimic, while at the same time remaining hydrolytically sta-ble. The synthesis of heteromeric dendritic structures<sup>[20](#page-11-0)</sup> has rarely been reported in the literature despite possible advantages of this architecture over conventional symmetric dendrimers. Asymmetrical dendrimers could be specifically targeted to various biomolecules, making it possible to develop multi-targeting molecular systems.

#### 2. Results and discussion

We have used the high chemoselectivity and bioorthogonality of the click reaction to design various peptide-based designer dendrimers. Amino acids with side chain functionality can be chosen as building blocks for dendrimer synthesis. The protective group chemistry of amino acids is already well established; we therefore chose to synthesize dendrimers based on amino acids. Our strategy was to anchor the alkyne on one dendron and the azide on the other dendron (Fig. 3a). These alkyne- and azide-truncated dendrons were then linked using click reactions to yield dendrimers. In



Fig. 3. Basic design principle for the synthesis of peptide dendrimers using click reaction.

another design strategy, a dialkyne core was chosen and the dendrons, functionalized with azide units, were coupled to yield the dendrimer (Fig. 3b). Alternately, diazido spacers were coupled to dendrons functionalized with alkyne units (Fig. 3c).

With this design strategy in mind, we synthesized alkyne- and azide-truncated peptide dendrons. Aspartic acid (Asp) is an attractive choice for the synthesis of peptide dendrons because the presence of a side chain carboxylic acid group allows it to act as a branching unit in dendrimer synthesis. Azide-truncated Aspbased dendrons 1a and 1b were synthesized in good yields following the strategy demonstrated in our earlier reports.<sup>18</sup>

Lysine (Lys) was chosen as the monomer unit to synthesize an alkyne-truncated dendron because it has a side chain amino group that can act as a branching unit. Different generations of alkyneterminated dendrons (2a, 2b, 3a, and 3b) were synthesized from BocLys(Boc)-OH.

With these dendrons (Fig. 4), we synthesized the dendrimers in a convergent fashion using the strategies outlined in Fig. 3. In order to synthesize asymmetrical dendrimers 4a, 4b, and 5, a dendron truncated with an alkyne was reacted with an azide-truncated dendron ([Scheme 1](#page-2-0)). To synthesize a symmetrical dendrimer, dendrons were attached to a symmetrical dialkyne or diazide core  $(Schemes 2–6)$  $(Schemes 2–6)$  $(Schemes 2–6)$  $(Schemes 2–6)$  $(Schemes 2–6)$ .



Fig. 4. Structures of various peptide-based dendrons.

In order to synthesize dual-surface dendrimers 4a, 4b, and 5, we linked an aspartic acid-based dendron appended with an azido group at the N-terminus and a lysine-based dendron truncated with an alkyne at the C-terminus [\(Scheme 1](#page-2-0)). The reaction of 1a and 2a in a Cu(I)-catalyzed 1,3-dipolar cycloaddition reaction produced 1,4-substituted triazole-based dendrimer  $4a$  in ~68% yield with four Boc protected amines and four carbomethoxy units. The Boc groups were deprotected to generate the cationic dendrimer 4b in quantitative yield ([Scheme 1](#page-2-0)). Dendrimer 4b was also synthesized

<span id="page-2-0"></span>

Scheme 1. Synthesis of unsymmetrical dendrimers 4a, 4b, and 5.

by reacting tetraamine 2b with 1a. Similarly, dendrimer 5 was synthesized from benzyl-protected Asp dendron 1b. Dendrimer 5 was synthesized to demonstrate that different groups can be added to the periphery of the dendritic surface. This robust and flexible dendrimer synthesis, able to generate amino terminals, carboxyl terminals, and a combination of the two, will be of profound interest to scientists working in various disciplines.

Cystine, with two carboxylate groups, can be functionalized with a propargyl amine to produce a dialkyne suitable as a core unit. The dialkyne core 7 was synthesized in 91% yield by coupling Boc-Cystine 6 with propargyl amine in the presence of DCC and N-hydroxysuccinimide (Scheme 2). Similarly, the cystine core 9, carrying two azide functionalities, was synthesized from cystine diOMe $\cdot$ 2HCl 8. The reaction between compound 8 and azidoacetyl chloride generated compound 9; this is an ideal core unit for a reaction with a Lys-dendron appended with an alkyne unit.

The reaction of Asp dendron 1a with 7 produced the symmetrical dendrimer 10a (Scheme 3). Similarly, the reaction of core 9 with dendron 2a afforded dendrimer 11a [\(Scheme 4\)](#page-3-0). Dendrimer 10a was deprotected with TFA to obtain 10b, with potential for further synthetic extrapolation. Lysine-based dendron 3a, on



Scheme 2. Synthesis of cystine-based cores (a) dialkyne core 7 (b) diazide core 9.



Scheme 3. Synthesis of dendrimers 10a and 10b.

reaction with azide core 9, provided dendrimer 12a. The yields of these reactions were in the range of  $65-92%$ . The post-reaction workup afforded reasonably pure compounds. Purified dendrimers were obtained by passing the compounds through small silica gel columns. The purity of all the dendrimers synthesized was confirmed by reversed phase analytical HPLC. The <sup>1</sup>H NMR spectrum of 11a showed the presence of a triazole CH, indicating the formation of the desired product. The high resolution mass spectrum (HRMS) of 11a (calculated mass 2136.1425, measured mass 2136.1436), further confirmed the formation of the proposed dendrimer.

A cystine core can be replaced with a p-xylylene or 1,4-diazidobutane. The reaction of an alkyne-terminated dendron (2a, 2b, 3a or 3b) with diazides (13 or 14) provided dendrimers 15a,15b,16a, 16b,17a,17b,18a, and 18b ([Schemes 5 and 6\)](#page-4-0). This chemoselectivity is an important aspect of the click reaction: it can be used to synthesize anionic and cationic peptide dendrimers in good yields.

It is desirable to have a dendrimeric structure with further possibilities for conjugation to biomolecular systems. In order to demonstrate the versatility of our strategy, we synthesized dendrimers using a Lys-Asp dendron as the central core unit. The dipeptide core 21 was synthesized from BocLys(Boc)-OH 19 and the dialkyne derivative of aspartic acid 20. This dipeptide core with two alkyne units can be attached to dendrons, producing a dendrimer. The Lys unit with  $\alpha$  and  $\epsilon$  amino groups can be deprotected for further chemical extrapolation. The Asp-based dendron 1a was attached via click reaction to the Lys-Asp scaffold 21 to give den-drimer 22 in 70% yield [\(Scheme 7](#page-6-0)). The unused Lys N $\alpha$  and N $\epsilon$ amino handles allow other dendrons to be attached. Moreover, these  $N\alpha$  and  $N\epsilon$  amino groups could in principle also be reacted with molecules of biological significance.

<span id="page-3-0"></span>

Scheme 4. Synthesis of symmetrical dendrimers 11a, 11b, 12a, and 12b.

It is noteworthy that the dendrimer 15a displayed gelling properties in a mixture of chloroform and hexane (1:2). Similarly, compound 21 gelated in a mixture of ethyl acetate and hexane (1:5). These gels were examined using scanning electron microscope (SEM) and were found to have a fibrous morphology [\(Fig. 5\)](#page-6-0). The fibers are having widths mostly in the range of 0.37 $-0.63$  µm. These observations suggest that the triazole-based dendrimers aggregate in solution and can be further fine-tuned to form organoor hydrogels. Gels have potential uses in pharmaceutical, cosmetics, new material development, and as sensors.[21](#page-11-0) Therefore, the gelation properties exhibited by compounds 15a and 21 open up new avenues for further exploration.

Peptide-carbohydrate conjugates are becoming immensely relevant because of their biological relevance.<sup>1,22</sup> Carbohydrate dendrimers have significant biological activity, for example, enhanced binding to cell surface receptors due to their multivalent nature.<sup>[23](#page-11-0)</sup> The multivalent effect is not fully understood and therefore requires an in-depth study. Dendrimers are excellent candidates to address the question of multivalency via the creation of diverse classes of dendrimers. We have demonstrated that our strategy is a versatile method for synthesizing dendrimers with various carbohydrate moieties on the periphery. The azide-functionalized carbohydrate unit was synthesized from the pentaacetylated carbohydrate derivatives. Reaction of peracetylated glucosyl acetate or galactosyl acetate with tin tetrachloride and trimethylsilyl azide provided the corresponding azides in good yield. $24$  The azide-functionalized carbohydrate was reacted with cores, such as 23 and 26 to yield dendrimers 25a,b and 27a,b ([Schemes 8 and 9\)](#page-6-0).

The topological and electronic similarities of a triazole ring to a peptide bond, the large dipole, and the presence of two nitrogens as hydrogen bond acceptors, make triazole an excellent peptide bond surrogate.[25](#page-11-0) A triazole unit may impart rigidity and protease stability to the dendritic structure. Therefore, incorporation of a non-hydrolysable heterocyclic moiety in the structure of a dendrimer may provide added advantages.

## 3. Conclusions

In conclusion, we have demonstrated a simple and elegant method to synthesize designer peptide dendrimers with complex architecture using click chemistry. A series of symmetric and asymmetric dendrimers were synthesized based on the 1,3-dipolar cycloaddition reaction. The yield of synthesis was good, ranging from 65 to 92%. The functional group-tolerant nature of the click reaction allowed us to link dendrons with unprotected

<span id="page-4-0"></span>

Scheme 5. Synthesis of symmetrical lysine dendrimers 15a, 15b, 16a, and 16b.

functionalities. The synthesis of dendrimers with a dipeptide (Lys-Asp) core is another feature of our design; the additional lysine can carry units important for biological recognition. We also found that dendrimer 15a and the core 21 are good organogelators. In summary, we have demonstrated a good design principle and synthesis of various peptide dendrimers with good yield and purity.

## 4. Experimental section

# 4.1. Preparation of lysine-based dendrons

4.1.1. Preparation of propargylamide of bis Boc-lysine. To an ice cold solution of BocLys(Boc)-OH (2.10 g, 6.07 mmol) in 90 ml of dry dichloromethane was added N-hydroxysuccinimide (0.766 g, 6.65 mmol), DCC (1.377 g, 6.67 mmol), and stirred for 10 min. Afterwards, propargyl amine (0.456 ml, 6.65 mmol) and triethylamine (0.926 ml, 6.65 mmol) were added and stirred overnight. Filtered the reaction mixture, washed the filtrate with  $0.2$  N H<sub>2</sub>SO<sub>4</sub>, water, and finally with saturated aqueous  $N$ aHCO<sub>3</sub> solution. The organic layer was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ , filtered, and evaporated to yield 2.20 g BocLys(Boc)-propargylamide.

Yield: 94%; [ $\alpha$ ] $_0^{23.4}$  – 1.8 (c 0.38, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.38–1.70 (s+m, 24H), 2.20 (t, J=2.4 Hz, 1H), 3.10 (m, 2H), 4.04 (m, 3H), 4.56 (br s,1H), 5.03 (br d,1H), 6.46 (br s,1H); IR (KBr) 3333, 2977, 2935, 2866, 1687, 1524, 1453, 1376, 1249, 1171  $\rm cm^{-1}$ ; HRMS calcd for  $C_{19}H_{34}N_3O_5$  m/z 384.2498, found m/z 384.2500.

4.1.2. Preparation of Lys dendron  $2a$ . To an ice cold solution of BocLys (Boc)-OH (1.982 g, 5.73 mmol) in 80 ml of dry dichloromethane was added N-hydroxysuccinimide (0.723 g, 6.28 mmol), DCC (1.297 g, 6.28 mmol) and stirred for 10 min. Afterwards, propargylamide of lysine (0.525 g, 2.87 mmol) and triethylamine (0.875 ml, 6.29 mmol) were added and stirred overnight. Filtered, washed the filtrate with  $0.2$  N H<sub>2</sub>SO<sub>4</sub>, water and finally with saturated aqueous NaHCO<sub>3</sub> solution. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated to yield 2.10 g of the compound 2a.

Yield: 87%;  $[\alpha]_D^{24.5}$  -45.0 (c 0.39, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.37-1.85 (s+m, 54H), 2.25 (s, 1H), 3.10 (m, 6H), 4.02 (s, 2H), 4.09-4.45 (m, 3H), 4.70-5.00 (m, 2H), 5.60 (br s, 1H), 5.95 (br s, 1H), 7.14 (br d, 2H), 7.43 (br s, 1H); IR (KBr) 3340, 2932, 2862, 1691, 1651, 1527, 1450, 1372, 1249, 1169 cm-1 ; HRMS calcd for  $C_{41}H_{73}N_{7}O_{11}$ Na *m*/z 862.5266, found *m*/z 862.5244.

4.1.3. Preparation of Lys dendron 3a. To an ice cold and well-stirred solution of BocLys(Boc)-OH (1.447 g, 4.18 mmol) in 70 ml of dry dichloromethane was added N-hydroxysuccinimide (0.481 g, 4.18 mmol), DCC (0.862 g, 4.18 mmol) and stirred for 10 min. To this was added propargylamide of Lys(Lys)Lys-OH (0.408 g, 0.929 mmol), triethylamine (0.581 ml, 4.18 mmol) and stirred overnight. Filtered, washed the filtrate with 0.2 N  $H<sub>2</sub>SO<sub>4</sub>$ , water and finally with saturated aqueous  $N$ aHCO<sub>3</sub> solution. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated to yield 1.22 g of the compound  $3a$ .

Yield: 75%;  $[\alpha]_D^{24}$  +28.66 (c 0.15, MeOH); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.15–2.05 (br m, 114H), 2.20 (s, 1H), 2.95–3.60 (br m, 14H), 3.90–4.60 (br m, 9H), 4.72–5.25 (br m, 4H), 5.52–6.18 (br m, 4H), 6.68-7.20 (br m, 2H), 7.32-7.90 (br m, 5H); IR (KBr) 3325, 3082, 2976, 2934, 2861, 2247, 1693, 1649, 1527, 1452, 1392, 1367, 1249, 1171 cm<sup>-1</sup>; HRMS calcd for C<sub>85</sub>H<sub>153</sub>N<sub>15</sub>O<sub>23</sub>Na m/z 1775.1161, found m/z 1775.1142.



Scheme 6. Synthesis of symmetrical Lys dendrimers based on aromatic and aliphatic cores.

4.1.4. Preparation of Lys dendron 3b. To an ice-cooled solution of 3a (0.160 g, 0.091 mmol) in dry  $CH_2Cl_2$  (1 ml) was added TFA (3 ml) and stirred at RT for 2 h. The reaction mixture was subjected to vacuum to afford 0.080 g of 3b.

Yield: 92%; <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz)  $\delta$  0.95–1.80 (m, 42H), 2.26  $(s, 1H)$ , 2.67 (br s, 8H), 2.87 (br s, 6H), 3.52-3.79 (m, 6H), 3.81-4.05  $(m, 3H)$ ; IR (KBr) 3430, 2935, 1664, 1436, 1383, 1197, 1138 cm<sup>-1</sup>; HRMS calcd for C<sub>45</sub>H<sub>90</sub>N<sub>15</sub>O<sub>7</sub> m/z 952.7148, found m/z 952.7155.

## 4.2. Preparation of azidoacetyl chloride

To an ice-cooled solution of sodium azide (0.936 g, 14.4 mmol) in 8 ml water was added bromoacetic acid (1.0 g, 7.19 mmol) and stirred for 24 h. The reaction mixture was acidified with HCl to pH  $\sim$  2 and extracted with diethyl ether (3 $\times$ 100 ml). The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, and evaporated in vacuo to afford 0.710 g of the product.

Yield: 97%; <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz)  $\delta$  3.72 (s, 2H); <sup>13</sup>C NMR (D2O, 75 MHz) d 52.58, 176.08; IR (KBr) 3449, 2950, 2128, 1612, 1409, 1299 cm $^{-1}$ .

To an ice cold solution of azidoacetic acid (0.404 g, 4.0 mmol) in 5 ml of dry dichloromethane was added 2 drops of dry DMF and oxalyl chloride (0.412 ml, 4.8 mmol) and stirred for 24 h and reaction mixture was evaporated to yield 0.450 g azidoacetyl chloride.

Yield: 94%; IR (KBr) 3375, 2928, 2109, 1835, 1746, 1642, 1419, 1284 cm $^{-1}$ .

## 4.3. Preparation of Z-Asp(AspdiOMe)AspdiOMe

To a well-stirred and ice-cooled solution of Z-Aspartic acid (1.50 g, 5.62 mmol) in 70 ml dry  $CH_2Cl_2$  was added N-hydroxysuucinimide (1.62 g, 14.07 mmol), DCC (2.90 g, 14.05 mmol), Asp·diOMe·HCl (2.77 g, 14.02 mmol) containing triethylamine (1.95 ml, 14.04 mmol). After stirring for 24 h at room temperature, the reaction mixture was filtered. The residue was washed with  $CH_2Cl_2$  (4×20 ml) and the combined filtrates were washed sequentially with 2 N  $H<sub>2</sub>SO<sub>4</sub>$ , water, and 5% aqueous NaHCO<sub>3</sub> solution. The organic layer was dried over  $Na<sub>2</sub>SO<sub>4</sub>$  and evaporated in vacuo to afford 2.90 g of the product.

<span id="page-6-0"></span>

Scheme 7. Synthesis of dendrimer with Lys-Asp as the central core unit.



**Fig. 5.** SEM images of gels from (a) 21 (Ethyl acetate/hexane  $(1:5)$ ) (b) **15a** (Chloroform/hexane (1:2)).



Scheme 8. Synthesis of various carbohydrate-based dendrimers on a trialkyne core.



Scheme 9. Synthesis of carbohydrate dendrimers on a hexaalkyne core.

Yield: 93%;  $[\alpha]_D^{24.4}$  –8.5 (c 0.26, MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  2.63-3.10 (m, 6H), 3.72 (s, 6H), 3.77 (s, 6H), 4.63 (m, 1H), 4.85 (m, 2H), 5.16 (s, 2H), 6.28 (d, J=6.6 Hz, 1H), 6.73 (d, J=8.1 Hz, 1H), 7.38 (s, 5H), 7.94 (br d, 1H); IR (KBr) 3307, 3084, 2955, 2854, 1739, 1698, 1648, 1545, 1438, 1366, 1300 cm $^{-1}$ ; HRMS calcd for C<sub>24</sub>H<sub>31</sub>N<sub>3</sub>O<sub>12</sub>Na *m*/z 576.1805, found *m*/z 576.1793.

## 4.4. Preparation of Asp dendron 1a

An ice-cooled solution of Z-Asp-(Asp $\cdot$ diOMe)Asp $\cdot$ diOMe (0.221 g, 0.4 mmol) in 10 ml of dry MeOH was admixed with 10% Pd/C, (peptide/catalyst 1:0.5 w/w), and  $H_2$  was bubbled through the reaction mixture for 1.5 h. After completion of the reaction (TLC), the solution was filtered, and the filtrate was evaporated. The residue obtained was dissolved in dry dichloromethane. The solution was cooled in an ice-bath and triethylamine (0.056 ml, 0.4 mmol) was added, followed by the slow addition of azidoacetyl chloride  $5(0.048 \text{ g}, 0.4 \text{ mmol})$  over a period of 0.5 h. The reaction mixture was left stirred at room temperature for 12 h. The solvent was removed in vacuo, the solid obtained was dissolved in ethyl acetate (50 ml), washed, with 2 N  $H<sub>2</sub>SO<sub>4</sub>$ , water, and 5% aqueous NaHCO<sub>3</sub> solution. The organic layer was dried over Na2SO4, evaporated and purified by silica gel column chromatography using EtOAc/hexane to give 0.144 g of 1a.

Yield: 72%;  $[\alpha]_D^{22.6}$  +54.7 (c 0.10, CHCl<sub>3</sub>); mp 135–136 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  2.60-3.20 (m, 6H), 3.70 (s, 6H), 3.75  $(s, 3H)$ , 3.76  $(s, 3H)$ , 4.02  $(s, 2H)$ , 4.82  $(m, 3H)$ , 6.85  $(d, J=7.4$  Hz, 1H), 7.65 (d, J=7.5 Hz, 1H), 7.87 (d, J=7.0 Hz, 1H); IR (KBr) 3310, 2959, 2854, 2111, 1741, 1666, 1644, 1536, 1436, 1373, 1281 cm $^{-1}$ ; HRMS calcd for C<sub>18</sub>H<sub>27</sub>N<sub>6</sub>O<sub>11</sub> m/z 503.1738, found m/z 503.1732.

## 4.5. Preparation of unsymmetrical dendrimer 4a

To an ice-cooled solution of 2a (0.151 g, 0.18 mmol) in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.031 ml, 0.18 mmol),  $N_3$ -Asp-(Asp $\cdot$ diOMe)Asp $\cdot$ diOMe 1a (0.090 g, 0.18 mmol) and CuI (0.004 g, 0.021 mmol). The reaction mixture was stirred under nitrogen atmosphere for 17 h. The reaction mixture was evaporated, the solid thus obtained was washed with 0.2 N H<sub>2</sub>SO<sub>4</sub>, water, NH<sub>4</sub>CI/NH<sub>4</sub>OH (9:1) solution and finally with water. The residue obtained was dried and crystallized from a mixture of chloroform and methanol to give 0.163 g dendrimer 4a.

Yield: 68%; [ $\alpha$ ] $_{\text{D}}^{24.6}$  – 9.4 (c 0.22, MeOH); mp 158–160 °C; <sup>1</sup>H NMR (DMSO- $d_6$ , 300 MHz)  $\delta$  1.10-1.70 (s+m, 54H), 2.71-3.20 (m, 12H), 3.60 (s, 12H), 3.81 (br s, 2H), 4.31 (m, 3H), 4.63 (m, 3H) 5.09 (br s, 2H), 6.74 (m, 2H), 6.90 (d, J=7.8 Hz, 1H), 7.74 (br s, 2H), 7.80 (s, 1H), 8.44 (m, 3H), 8.58 (d, J=7.8 Hz, 2H); IR (KBr) 3438, 2932, 1753, 1643, 1533, 1448, 1370, 1246, 1171 cm<sup>-1</sup>; HRMS calcd for C<sub>59</sub>H<sub>99</sub>N<sub>13</sub>O<sub>22</sub>Na m/z 1364.6925, found m/z 1364.6909.

## 4.6. Preparation of unsymmetrical dendrimer 4b

To an ice-cooled solution of  $4a$  (0.025 g, 0.0186 mmol) in 1 ml of  $\text{dry } CH_2Cl_2$  was added 3 ml of TFA. The solution was left stirred for 2 h and the reaction was monitored by TLC. The reaction mixture was subjected to high vacuum to yield 0.016 g of the dendrimer 4b.

Yield: 91%; <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz)  $\delta$  1.00–2.00 (m, 18H),  $2.55-3.20$  (m, 12H), 3.62 (m, 12H), 3.80 $-4.00$  (m, 3H), 4.00 $-4.20$ (m, 2H), 4.67 (m, 2H, merged with  $D_2O$  residual peak), 5.10–5.30 (m, 3H), 7.83 (s, 1H); HRMS calcd for  $C_{39}H_{68}N_{13}O_{14}$  m/z 942.5009, found m/z 942.5039.

## 4.7. Preparation of Dendrimer 5

4.7.1. Preparation of Boc-Asp(OBzl)OBzl. To a solution of Boc-Asp-OH (3.5 g, 15.01 mmol) in 50 mL dry dichloromethane was added Nhydroxysuccinimide (3.7 g, 32.14 mmol), DCC (6.6 g, 32.14 mmol), benzyl alcohol (3.3 ml, 31.7 mmol), and triethylamine (4.4 ml, 32.14 mmol). The reaction mixture was kept at  $0^{\circ}$ C and stirred for 12 h. Evaporated the reaction mixture, dissolved in ethyl acetate, and washed with 0.2 N  $H<sub>2</sub>SO<sub>4</sub>$  and water. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated to obtain 4.58 g of Boc-Asp(OBzl)OBzl.

Yield 73.9%;  $[\alpha]_D^{25.8}$  +4.90 (c 0.10, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.44 (s, 9H), 2.88 (dd, J=17.4 Hz, 4.8 Hz, 1H), 3.07 (dd, J=17.4 Hz, 4.8 Hz, 1H), 4.65 (m, 1H), 5.08 (s, 2H), 5.14 (s, 2H), 5.53 (d, J=7.8 Hz, 1H), 7.27–7.42 (m, 10H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz) δ 28.5, 37.1, 50.3, 67.0, 67.6, 80.4, 128.6, 128.8, 135.6, 155.6, 171.1; IR (KBr) 3433, 3032, 2975, 1742 (br), 1504, 1457, 1355, 1165 cm<sup>-1</sup>; HRMS calcd for C<sub>23</sub>H<sub>27</sub>NO<sub>6</sub>Na  $m/z$  436.1736, found  $m/z$  436.1738.

4.7.2. Preparation of Boc-Asp(Asp(OBzl)OBzl)-Asp(OBzl)OBzl. To an ice-cooled and well-stirred solution of Boc-Asp-OH (1.02 g, 4.4 mmol) in 50 mL of dry dichloromethane was added Nhydroxysuccinimide (1.12 g, 9.6 mmol) and DCC (2.0 g, 9.6 mmol). The reaction mixture was stirred for 10 min, added Asp(OBzl)OBzl (3.0 g, 9.6 mmol) and triethylamine (1.3 ml, 9.6 mmol). The reaction mixture was stirred for 12 h at room temperature and filtered. Filtrate was washed sequentially with aqueous 0.2 N  $H<sub>2</sub>SO<sub>4</sub>$ ,  $NaHCO<sub>3</sub>$ , and water. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated to get 2.48 g of Boc-Asp(Asp(OBzl) OBzl)-Asp(OBzl)OBzl.

Yield: 67.3%;  $[\alpha]_D^{23}$  +33.16 (c 0.39, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) δ 1.44 (s, 9H), 2.51 (m, 1H), 2.84 (dd, J=17.0, 4.6 Hz, 2H), 2.98  $(dd, J=17.1, 4.8$  Hz, 2H), 4.48 (m, 1H), 4.82 (m, 2H), 5.03 (s, 2H), 5.05  $(s, 2H)$ , 5.08  $(s, 2H)$ , 5.11  $(s, 2H)$ , 5.97  $(d, J=6.0$  Hz, 1H), 6.75  $(d, J=8.1$  Hz, 1H), 7.20–7.40 (br m, 20H), 7.60 (d, J=7.5 Hz, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz) d 28.2, 36.1, 37.4, 48.8, 50.9, 66.8, 67.5, 80.2, 128.3, 128.5, 135.1,135.3,155.5,170.2,170.6,170.9; IR (KBr) 3319, 3064, 3034, 2975, 1738, 1696, 1650, 1529, 1455, 1388, 1361, 1284, 1169 cm<sup>-1</sup>; HRMS calcd for C<sub>45</sub>H<sub>49</sub>N<sub>3</sub>O<sub>12</sub>Na *m*/z 846.3214, found *m*/z 846.3247.

#### 4.8. Preparation of azide-truncated dendron 1b

To an ice-cooled solution of Asp(Asp(OBzl)OBzl)-Asp(OBzl)OBzl (0.263 g, 0.36 mmol) in 30 ml of dry dichlorormethane was added triethylamine (0.25 ml, 1.8 mmol) and followed by addition of azidoacetyl chloride (0.065 g, 0.54 mmol). The reaction mixture was stirred for 12 h and washed sequentially with 2 N  $H<sub>2</sub>SO<sub>4</sub>$ , NaHCO<sub>3</sub> solution, and water. The organic layer was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$  and evaporated to get 0.170 g of dendron.

Yield: 58%;  $[\alpha]_D^{21.4}$  +18.1 (c 0.09, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  2.47-2.62 (m, 1H), 2.70-3.10 (m, 5H), 3.91 (s, 2H), 4.71-4.92 (m, 3H), 4.99-5.17 (m, 8H), 6.83 (d,  $J=7.8$  Hz, 1H), 7.21-7.38 (m, 20H), 7.65 (d, J=7.2 Hz, 1H), 7.81 (d, J=7.2 Hz, 1H); IR  $(KBr)$  3428, 2925, 2855, 2108, 1738, 1651, 1540, 1390, 1182 cm<sup>-1</sup>; ESI  $MS m/z 807 (M+H<sup>+</sup>).$ 

## 4.9. Preparation of dendrimer 5

To an ice-cooled solution of 2a (0.042 g, 0.050 mmol) in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.008 ml, 0.05 mmol), 1b (0.040 g, 0.05 mmol) and CuI (0.0093 g, 0.005 mmol). The reaction mixture was stirred under nitrogen atmosphere for 17 h. The reaction mixture was evaporated, the solid thus obtained was washed with  $0.2$  N  $H<sub>2</sub>SO<sub>4</sub>$ , water, NH4CI/NH4OH (9:1) solution and finally with water. The residue obtained was dried and crystallized from a mixture of chloroform and methanol to give 0.0524 g of dendrimer 5.

Yield: 64%; [ $\alpha$ ] $_{D}^{25.1}$  –8.7 (c 0.24, CHCl<sub>3</sub>+MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.20-1.80 (m, 54H), 2.49-3.30 (m, 12H), 3.90-4.29 (br m, 2H),  $4.29-4.40$  (br m, 2H),  $4.51$  (br m, 2H),  $4.70-4.90$  (m, 4H),  $4.95-5.21$  (m, 8H),  $5.40-5.85$  (br d, 3H),  $5.92$  (br s, 2H),  $6.70-7.10$ (br d, 2H), 7.20-7.50 (br m, 20H), 7.50-7.70 (br m, 2H), 7.77-7.90 (br m, 2H); IR (KBr) 3311, 3069, 2928, 2859, 1688, 1646, 1529, 1456, 1387, 1248, 1171 cm<sup>-1</sup>; ESI MS found m/z 1668 (M+Na<sup>+</sup>).

## 4.10. Preparation of cystine core 7

To an ice cold solution of bis Boc-Cystine (0.8 g, 1.82 mmol) in 90 ml of dry dichloromethane was added N-hydroxysuccinimide (0.460 g, 3.99 mmol) and DCC (0.824 g, 3.99 mmol). The reaction mixture was stirred for 10 min, afterwards propargyl amine (0.274 ml, 3.99 mmol) and triethylamine (0.56 ml, 4.02 mmol) were added and stirred overnight. Filtered the reaction mixture, washed the filtrate with 0.2 N  $H<sub>2</sub>SO<sub>4</sub>$ , water, and finally with saturated aqueous NaHCO<sub>3</sub> solution. The organic layer was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ , filtered, and evaporated to yield 0.850 g of the compound 7.

Yield: 91%;  $[\alpha]_D^{24.9}$  +57.7 (c 0.38, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) d 1.48 (s, 18H), 2.18 (s, 2H), 2.91 (m, 4H), 4.25 (m, 4H), 4.93 (br s, 2H), 5.55 (d, J=9.3 Hz, 2H), 8.06 (br s, 2H); IR (KBr) 3307, 2978, 2928, 2853, 1660, 1525, 1446, 1372, 1251, 1168 cm $^{-1}$ ; HRMS calcd for C<sub>22</sub>H<sub>34</sub>N<sub>4</sub>O<sub>6</sub>S<sub>2</sub>Na m/z 537.1817, found m/z.

#### 4.11. Preparation of cystine core 9

To an ice-cooled solution of Cystine diOMe $\cdot$ 2HCl (0.674 g, 1.976 mmol) in 40 ml dry dichloromethane was added dry triethylamine (1.10 ml, 7.91 mmol) followed by the slow addition of azidoacetyl chloride (0.473 g, 3.96 mmol) over a period of 0.5 h and stirred at room temperature for 12 h. The solvent was removed in vacuo, and the residue was dissolved in ethyl acetate (50 ml), washed with 2 N  $H_2SO_4$ , water, and 5% aqueous NaHCO<sub>3</sub> solution. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated to yield 0.670 g of the compound 9.

Yield: 78%;  $[\alpha]_D^{23.4}$  +58.7 (c 0.23, CHCl<sub>3</sub>); <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz) d 3.06 (m, 2H), 3.33 (m, 2H), 3.76 (s, 6H), 4.06 (s, 4H), 4.83 (m, 2H); IR (KBr) 3331, 3069, 2996, 2953, 2102, 1731, 1658, 1546, 1442, 1394, 1326, 1240 cm<sup>-1</sup>; HRMS calcd for C<sub>12</sub>H<sub>18</sub>N<sub>8</sub>O<sub>6</sub>S<sub>2</sub>K m/z 473.0428, found m/z 473.0439.

## 4.12. Preparation of symmetrical Asp dendrimer 10a

To an ice-cooled solution of dialkyne 7 (0.52 g, 0.101 mmol) in 20 ml of dry acetonitrile under nitrogen was added diisopropylethylamine (0.037 ml, 0.215 mmol),  $N_3$ -Asp-(Asp $\cdot$ diOMe) Asp diOMe 1a (0.101 g, 0.201 mmol), and CuI (0.004 g, 0.021 mmol). The reaction mixture was stirred under  $N_2$  atmosphere for 17 h and the solvent was evaporated. The solid thus obtained was dissolved in chloroform, washed with 2 N  $H<sub>2</sub>SO<sub>4</sub>$ , water, NH4CI/NH4OH (9:1) solution and finally with water. The organic layer was dried over anhydrous Na2SO4, filtered, and evaporated to yield 0.10 g of the dendrimer 10a.

Yield: 65%; [ $\alpha$ ] $_{D}^{25.5}$  +23.4 (c 0.28, CHCl<sub>3</sub>); <sup>1</sup>H NMR (DMSO- $d_6$ , 300 MHz)  $\delta$  1.35 (s, 18H), 2.65-2.90 (m, 12H), 3.00-3.25 (m, 4H),  $3.60$  (s, 24H),  $4.00 - 5.00$  (m, 6H),  $4.61$  (s, 6H),  $5.07$  (s, 4H),  $7.02$  (s, 2H) 7.81 (s, 2H), 8.37-8.70 (m, 8H); IR (KBr) 3321, 2958, 1739, 1654, 1537, 1444, 1370, 1231, 1173 cm-1 ; HRMS calcd for  $C_{58}H_{86}N_{16}O_{28}S_2N$ a *m/z* 1541.5137, found *m/z* 1541.5132.

#### 4.13. Preparation of symmetrical Asp dendrimer 10b

To an ice-cooled solution of 10a (0.050 g, 0.033 mmol) in 1 ml of dry CH2Cl2 was added 2 ml of TFA. The solution was left stirred at  $0 °C$  for 2 h and the reaction mixture was subjected to high vacuum to yield 0.040 g of the dendrimer 10b.

Yield: 92%;  $^1$ H NMR (DMSO- $d_6$ , 300 MHz)  $\delta$  1.23 (m, 4H),  $2.62 - 2.75$  (m, 12H),  $2.80 - 3.08$  (m, 4H),  $3.62$  (s, 24H),  $4.07$  (m, 2H), 4.36 (m, 2H), 4.47-4.80 (m, 8H), 5.11 (s, 4H), 7.93 (s, 2H), 8.22-8.72

 $(m, 6H)$ , 9.13 (s, 2H); <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz)  $\delta$  2.55–2.90 (m, 12H),  $2.95-3.18$  (m, 4H), 3.51 (s, 6H), 3.56 (s, 6H), 3.59 (s, 6H), 3.61 (s, 6H), 4.15e4.55 (m, 8H), 5.16 (s, 8H), 7.99 (s, 2H); IR (KBr) 3259, 3071, 2962, 1738, 1669, 1541, 1440, 1375, 1177 cm<sup>-1</sup>; HRMS calcd for  $C_{48}H_{70}N_{16}O_{24}S_2$ Na *m*/z 1341.4088, found *m*/z 1341.4077.

## 4.14. Preparation of symmetrical Lys dendrimer 11a

To an ice-cooled solution of  $2a(0.240 g, 0.286 mmol)$  in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.05 ml, 0.29 mmol), 9 (0.62 g, 0.142 mmol), and CuI (0.006 g, 0.031 mmol). The reaction mixture was stirred under  $N_2$ atmosphere for 18 h. The reaction mixture was evaporated, dissolved in ethyl acetate. Washed the filtrate with  $0.2$  N  $H_2SO_4$  water,  $NH<sub>4</sub>Cl/NH<sub>4</sub>OH$  (9:1) solution, and finally with water. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated to yield 0.440 g of the dendrimer 11a.

Yield: 73%;  $[\alpha]_D^{20.9}$  –16.8 (c 0.26, MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.27-1.82 (m, 108H), 2.87-3.46 (m, 12H), 3.68 (s, 6H),  $3.87 - 4.67$  (m, 12H),  $4.78$  (m, 6H),  $5.08$  (m, 4H),  $5.37 - 6.00$  (m, 4H), 6.70-7.17 (m, 4H), 7.35-7.80 (m, 6H), 7.94 (s, 2H); IR (KBr) 3325, 2975, 2935, 1690 (br), 1523, 1451, 1367, 1248, 1171 cm<sup>-1</sup>; HRMS calcd for C<sub>94</sub>H<sub>164</sub>N<sub>22</sub>O<sub>28</sub>S<sub>2</sub>Na m/z 2136.1425, found m/z 2136.1436.

# 4.15. Preparation of symmetrical Lys dendrimer 11b

To an ice-cooled solution of 11a (0.030 g, 0.0142 mmol) in 0.5 ml of dry CH2Cl2 was added 2 ml of TFA. The solution was left stirred for 2 h and subjected to high vacuum to yield dendrimer 11b.

Yield: 91%; IR (KBr) 3403, 2927, 1744, 1682, 1542, 1439, 1180 cm<sup>-1</sup>; HRMS calcd for C<sub>54</sub>H<sub>101</sub>N<sub>22</sub>O<sub>12</sub>S<sub>2</sub> m/z 1313.7411, found m/z 1313.7411.

#### 4.16. Preparation of symmetrical Lys dendrimer 12a

To an ice-cooled solution of Lys<sub>7</sub>-alkyne 3a (0.30 g, 0.171 mmol) in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.03 ml, 0.174 mmol), 9 (0.0373 g, 0.086 mmol), and CuI (0.006 g, 0.031 mmol). The reaction mixture was stirred under  $N_2$  atmosphere for 17 h and solvent was evaporated. The solid thus obtained was dissolved in ethyl acetate, washed with 0.2 N  $H_2SO_4$ , water, NH<sub>4</sub>CI/NH<sub>4</sub>OH (9:1) solution, and finally with water. The organic layer was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ , filtered, and evaporated to yield dendrimer 12a.

Yield: 67%;  $[\alpha]_D^{21.5}$  –24.1 (c 0.17, MeOH); <sup>1</sup>H NMR (DMSO- $d_6$ , 300 MHz)  $\delta$  0.90-1.60 (m, 228H), 2.60-2.70 (m, 28H), 3.43 (m, 10H), 3.63 (m, 10H), 3.87-4.20 (m, 12H), 4.41 (m, 2H), 4.89 (m, 4H), 6.08 (m, 10H), 6.28 (m, 4H), 7.15-7.42 (br m, 10H), 7.59 (br d, 2H), 7.83 (br d, 2H), 8.49 (br d, 2H); IR (KBr) 3299, 2930, 2861, 1691, 1646, 1527, 1453, 1367, 1248, 1171 cm<sup>-1</sup>; HRMS calcd for  $C_{182}H_{324}N_{38}O_{52}S_2N$ a *m/z* 3961.3216, found *m/z* 3961.3030.

#### 4.17. Preparation of symmetrical Lys dendrimer 12b

To an ice-cooled solution of 12a (0.040 g, 0.010 mmol) in 0.5 ml of dry CH2Cl2 was added 3 ml of TFA. The solution was left stirred for 2 h and subjected to high vacuum to yield dendrimer **12b**.

Yield: 92%; <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz)  $\delta$  1.07–1.90 (m, 84H),  $2.75-3.20$  (m, 28H), 3.62 (s, 6H), 3.70-4.22 (m, 18H), 4.25-4.45 (m, 6H), 5.19 (m, 4H), 7.80 (s, 2H); IR (KBr) 3433, 3089, 2929, 1677, 1544, 1435, 1201, 1135 cm<sup>-1</sup>; HRMS calcd for C<sub>102</sub>H<sub>197</sub>N<sub>38</sub>O<sub>20</sub>S<sub>2</sub> m/z 2338.5008, found m/z 2338.5006.

## 4.18. Preparation of symmetrical dendrimer 15a

To an ice-cooled solution of 2a (0.420 g, 0.50 mmol) in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.14 ml, 0.81 mmol), p-xylylene diazide (13) (0.047 g, 0.25 mmol), and CuI (0.010 g, 0.053 mmol). The reaction mixture was stirred under  $N_2$  atmosphere for 18 h, evaporated the solvent, and dissolved in ethyl acetate. It was then washed with 0.2 N H<sub>2</sub>SO<sub>4</sub>, water, NH<sub>4</sub>Cl/NH<sub>4</sub>OH (9:1) solution, and finally with water. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated to yield 0.377 g of the dendrimer 15a.

Yield: 81%;  $[\alpha]_D^{23.8}$  –13.5 (c 0.57, MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.10-1.90 (s+m, 108H), 2.70-2.95 (br s, 12H), 3.80-4.12 (br m, 4H),  $4.12-4.45$  (m, 6H),  $4.65$  (br s, 2H), 5.0 (br s, 2H),  $5.15-5.40$  (m, 4H),  $5.72$  (br s, 4H),  $6.95-7.25$  (m, 8H),  $7.36$  (s, 2H), 7.50 (s, 2H); IR (KBr) 3326, 2931, 2858, 1700, 1659, 1530, 1452, 1367, 1249, 1171 cm<sup>-1</sup>; HRMS calcd for C<sub>90</sub>H<sub>154</sub>N<sub>20</sub>O<sub>22</sub>Na m/z 1890.1444, found m/z 1890.1463.

## 4.19. Preparation of symmetrical dendrimer 15b

To an ice-cooled solution of 15a (0.101 g, 0.054 mmol) in 0.5 ml of dry CH2Cl2 was added 2 ml of TFA. The solution was left stirred for 2 h and subjected to high vacuum to yield 0.057 g of the dendrimer 15b.

Yield: 99%; <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz)  $\delta$  1.00–1.85 (br m, 36H), 2.25-3.25 (br m, 12H), 3.72-3.90 (m, 6H), 4.11 (s, 2H), 4.20-4.45  $(m, 4H)$ , 5.48 (s, 2H), 7.10–7.30 (br s, 4H), 7.85 (s, 2H); IR (KBr) 3419, 2928, 1674, 1548, 1431, 1198 cm $^{-1}$ ; HRMS calcd for  $\mathsf{C}_{50}\mathsf{H}_{91}\mathsf{N}_{20}\mathsf{O}_6$  m/z 1067.7430, found m/z 1067.7409.

## 4.20. Preparation of symmetrical dendrimer 16a

To an ice-cooled solution of 2a (0.420 g, 0.50 mmol) in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.14 ml, 0.81 mmol), butyl diazide (14) (0.035 g, 0.25 mmol), and CuI (0.010 g, 0.053 mmol). The reaction mixture was stirred under  $N_2$  atmosphere for 18 h. The reaction mixture was evaporated and dissolved in ethyl acetate. It was then washed with 0.2 N H<sub>2</sub>SO<sub>4</sub>, water, NH<sub>4</sub>Cl/NH<sub>4</sub>OH (9:1) solution, and finally with water. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated to yield 0.354 g of the dendrimer 16a.

Yield: 78%;  $[\alpha]_D^{26.3}$  –17.4 (c 0.38, MeOH); <sup>1</sup>H NMR (DMSO- $d_6$ , 300 MHz)  $\delta$  1.15-1.81 (s+m, 112H), 2.82-3.12 (br m, 12H), 3.19 (s, 4H), 3.78-3.92 (m, 4H), 4.18-4.41 (m, 6H), 6.65-6.78 (m, 6H), 6.83-6.95 (br d, 2H), 7.70-7.80 (br m, 4H), 7.86 (br s, 2H), 8.38 (br m, 2H); IR (KBr) 3328, 3082, 2976, 2935, 2866, 1694, 1650, 1522, 1453, 1367, 1248, 1171 cm<sup>-1</sup>; HRMS calcd for C<sub>86</sub>H<sub>154</sub>N<sub>20</sub>O<sub>22</sub>Na m/z 1842.1444, found m/z 1842.1384.

## 4.21. Preparation of symmetrical dendrimer 16b

To an ice-cooled solution of 16a (0.101 g, 0.056 mmol) in 0.5 ml of dry  $CH_2Cl_2$  was added 2 ml of TFA. The solution was left stirred for 2 h and subjected to high vacuum to yield 0.056 g of the dendrimer 16b.

Yield: 98%; <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz)  $\delta$  1.15–2.02 (br m, 40H),  $2.25-3.25$  (br m, 12H),  $3.68-4.21$  (m, 8H),  $4.27-4.50$  (m, 6H), 7.80-7.91 (m, 2H); IR (KBr) 3079, 2938, 1674, 1546, 1432, 1198 cm<sup>-1</sup>; HRMS calcd for C<sub>46</sub>H<sub>91</sub>N<sub>20</sub>O<sub>6</sub> m/z 1019.7430, found m/z 1019.7403.

## 4.22. Preparation of symmetrical dendrimer 17a

To an ice-cooled solution of  $3a(0.430 g, 0.25 mmol)$  in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.071 ml, 0.41 mmol), 13 (0.024 g, 0.128 mmol), and CuI (0.005 g, 0.026 mmol). The reaction mixture was stirred under  $N_2$  atmosphere for 17 h and solvent was evaporated. The solid thus obtained was dissolved in ethyl acetate, washed with 0.2 N  $H_2SO_4$  water, NH<sub>4</sub>Cl/NH<sub>4</sub>OH (9:1) solution, and finally with water. The organic layer was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ , filtered, and evaporated to yield 0.321 g of the dendrimer 17a.

Yield: 68%; <sup>1</sup>H NMR (DMSO- $d_6$ , 300 MHz)  $\delta$  1.00–1.80 (m, 228H),  $2.85-3.15$  (m, 28H), 3.80 $-4.00$  (m, 18H), 4.12 $-4.40$  (br m, 10H), 5.64 (br s, 4H), 6.62-7.05 (br m, 12H), 7.65-8.05 (br m, 12H), 8.36 (br s, 2H); IR (KBr) 3329, 2930, 2855, 1693, 1632, 1575, 1532, 1448, 1367, 1247, 1171  $\text{cm}^{-1}$ ; HRMS calcd for C<sub>178</sub>H<sub>314</sub>N<sub>36</sub>O<sub>46</sub>Na m/z 3715.3236, found m/z 3715.3226.

#### 4.23. Preparation of symmetrical dendrimer 17b

To an ice-cooled solution of 17a (0.101 g, 0.027 mmol) in 0.5 ml of dry  $CH<sub>2</sub>Cl<sub>2</sub>$  was added 3 ml of TFA. The solution was left stirred for 2 h and subjected to high vacuum to yield 0.057 g of the dendrimer 17b.

Yield: 100%; <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz)  $\delta$  1.00–1.85 (m, 84H),  $2.80-3.10$  (m,  $28H$ ),  $3.62-3.92$  (m,  $6H$ ),  $4.02-4.20$  (m,  $10H$ ), 4.26-4.50 (br m, 2H), 5.47 (s, 4H), 7.19 (s, 4H), 7.89 (br s, 2H); IR (KBr) 3435, 2575, 1779, 1670, 1340, 1171 cm<sup>-1</sup>; HRMS calcd for  $C_{98}H_{187}N_{36}O_{14}$  m/z 2092.5028, found m/z 2092.4950.

# 4.24. Preparation of symmetrical dendrimer 18a

To an ice-cooled solution of 3a (0.438 g, 0.25 mmol) in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.071 ml, 0.41 mmol), 14 (0.0175 g, 0.125 mmol), and CuI (0.005 g, 0.026 mmol). The reaction mixture was stirred under  $N_2$  atmosphere for 17 h and solvent was evaporated. The solid thus obtained was dissolved in ethyl acetate, washed with 0.2 N  $H_2SO_4$  water, NH<sub>4</sub>Cl/NH<sub>4</sub>OH (9:1) solution, and finally with water. The organic layer was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ , filtered, and evaporated to yield 0.388 g of the dendrimer 18a.

Yield: 85%;  $[\alpha]_D^{26.2}$  –11.2 (c 0.25, MeOH); <sup>1</sup>H NMR (DMSO- $d_6$ , 300 MHz)  $\delta$  1.13-1.82 (m, 232H), 2.80-3.10 (m, 28H), 3.86 (br m, 12H), 4.11-4.41 (m, 12H), 5.55-5.69 (br m, 4H), 6.60-6.90 (m, 12H), 7.60-8.01 (br m, 12H), 8.32 (br s, 2H); IR (KBr) 3332, 2931, 2856, 1693, 1639, 1529, 1450, 1393, 1367, 1247, 1171 cm<sup>-1</sup>; HRMS calcd for  $C_{174}H_{314}N_{36}O_{46}$ Na *m*/*z* 3667.3236, found *m*/*z* 3667.3271.

## 4.25. Preparation of symmetrical dendrimer 18b

To an ice-cooled solution of 18a (0.040 g, 0.011 mmol) in 0.5 ml of dry  $CH_2Cl_2$  was added 3 ml of TFA. The solution was left stirred for 2 h and subjected to high vacuum to yield 0.022 g of the dendrimer 18b.

Yield: 100%; <sup>1</sup>H NMR (D<sub>2</sub>O, 300 MHz)  $\delta$  1.15–1.90 (m, 88H),  $2.85-3.20$  (m,  $28H$ ),  $3.80-4.00$  (m,  $6H$ ),  $4.08-4.40$  (m,  $6H$ ),  $4.70$  (m, 10H) 8.36 (br s, 2H); IR (KBr) 3324, 3071, 2929, 2854, 1700, 1626, 1579, 1444, 1313, 1182 cm<sup>-1</sup>; HRMS calcd for C<sub>94</sub>H<sub>187</sub>N<sub>36</sub>O<sub>14</sub> m/z 2044.5028, found m/z 2044.5069.

#### 4.26. Preparation of Lys-Asp core 21

4.26.1. Preparation of 20. To an ice cold solution of Boc-Asp-OH (0.8 g, 3.43 mmol) in 90 ml of dry dichloromethane was added N-hydroxysuccinimide (0.869 g, 7.55 mmol) and DCC (1.557 g, 7.55 mmol) and stirred for 10 min. To this propargyl amine (0.517 ml, 7.54 mmol) and triethylamine (1.050 ml, 7.55 mmol) were added and stirred overnight. Filtered, washed the filtrate with 0.2 N  $H_2SO_4$ , water, and finally with saturated aqueous NaHCO<sub>3</sub> solution. The organic layer was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ ,

filtered, and evaporated to yield 0.90 g of the compound Boc-Aspdialkyne.

Yield: 85%; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) δ 1.45 (m, 9H), 2.23 (m, 2H), 2.57 (m, 1H), 2.89 (m, 1H), 4.04 (m, 4H), 4.48 (m, 1H), 6.14 (br s, 1H), 6.39 (br s, 1H), 7.19 (br s, 1H); IR (KBr) 3312, 2932, 2127, 1694, 1643, 1532, 1443, 1332, 1276, 1167 cm<sup>-1</sup>; HRMS calcd for  $C_{15}H_{21}N_3O_4$ Na *m*/z 330.1430, found *m*/z 330.1429.

To an ice-cooled solution of Boc-Asp-dialkyne (0.700 g, 2.28 mmol) in 6 ml of dry  $CH<sub>2</sub>Cl<sub>2</sub>$  was added 3 ml of TFA. The solution was left stirred for 2 h and completion of the reaction was monitored by TLC. The reaction mixture was subjected to high vacuum to yield 0.472 g of the NH2-Asp-dialkyne 20. It was used as such without further purification.

4.26.2. Preparation of 21. BocLys(Boc)-OH (0.788 g, 2.28 mmol) was dissolved in 90 ml of dry dichloromethane, cooled to 0  $\degree$ C. To this ice cold solution was added N-hydroxysuccinimide (0.262 g, 2.27 mmol), DCC (0.469 g, 2.27 mmol), and stirred for 10 min. Afterwards  $NH_2$ -Asp-dialkyne 20 (0.472 g, 2.28 mmol) and triethylamine (1.268 ml, 9.11 mmol) were added. The reaction mixture was stirred overnight and filtered. Washed the filtrate with 0.2 N  $H<sub>2</sub>SO<sub>4</sub>$ , water, and finally with saturated aqueous NaHCO<sub>3</sub> solution. The organic layer was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ , filtered, and evaporated to yield compound 21.

Yield: 80%;  $[\alpha]_D^{23.9}$  +1.8 (c 0.50, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.35-2.00 (s+m, 24H), 2.20 (m, 2H), 2.55 (m, 1H), 2.95 (m, 1H), 3.12 (m, 2H), 4.01 (m, 4H), 4.73 (br s, 2H), 5.61 (br s, 1H), 6.68 (br s, 1H), 7.27 (br s, 1H), 7.73 (br s, 1H), 8.04 (br s, 1H); IR (KBr) 3289, 2977, 2933, 2110, 1689, 1647, 1529, 1447, 1367, 1250, 1172 cm<sup>-1</sup>; HRMS calcd for C<sub>26</sub>H<sub>41</sub>N<sub>5</sub>O<sub>7</sub>Na *m*/z 558.2904, found *m*/z 558.2904.

## 4.27. Preparation of Asp dendrimer 22 with Lys-Asp as the central core

To an ice-cooled solution of 21 (0.050 g, 0.093 mmol) in 20 ml of dry acetonitrile under nitrogen was added diisopropylethylamine (0.016 ml, 0.093 mmol),  $N_3$ -Asp-(Asp $\cdot$ diOMe)Asp $\cdot$ diOMe 1a (0.094 g, 0.187 mmol), and CuI (0.004 g, 0.021 mmol). The reaction mixture was stirred under  $N_2$  atmosphere for 18 h. The reaction mixture was evaporated, dissolved in chloroform, and filtered. Washed the filtrate with 0.2 N H<sub>2</sub>SO<sub>4</sub>, water, NH<sub>4</sub>CI/NH<sub>4</sub>OH (9:1) solution, and finally with water. The organic layer was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ , filtered, and evaporated to yield 0.10 g of the dendrimer 22.

Yield: 70%; <sup>1</sup>H NMR (DMSO- $d_6$ , 300 MHz)  $\delta$  1.10–1.60 (s+m, 24H), 2.60-2.93 (m, 16H), 3.61 (s, 24H), 3.82 (m, 2H), 4.29 (m, 4H), 4.48e4.72 (m, 6H), 5.07 (m, 4H), 6.71 (br s, 1H), 6.94 (br s, 1H), 7.75-7.90 (m, 2H), 7.97-8.20 (m, 2H), 8.31-8.60 (m, 7H); IR (KBr) 3290, 3081, 2951, 1739, 1648, 1541, 1442, 1368, 1236, 1173 cm $^{-1}$ ; HRMS calcd for  $C_{62}H_{93}N_{17}O_{29}Na$   $m/z$  1562.6223, found  $m/z$ 1562.6230.

## 4.28. Preparation of carbohydrate dendrimers

4.28.1. Synthesis of dendrimers  $25a-b$ . To an ice-cooled and wellstirred solution of trialkyne 23 (0.032 g, 0.10 mmol) in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.06 ml, 0.35 mmol), azido sugar (24a or 24b) (0.131 g, 0.35 mmol), and CuI (0.007 g, 0.035 mmol). The reaction mixture was stirred under  $N_2$  atmosphere for 17 h and solvent was evaporated. The solid thus obtained was dissolved in ethyl acetate, washed with 0.2 N  $H_2SO_4$ , water,  $NH_4Cl/NH_4OH$  (9:1) solution, and finally with water. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated to yield the dendrimer.

4.28.1.1. Dendrimer **25a**. Yield: 87%;  $[\alpha]_D^{27.1}$  -16.3 (c 0.08, CHCl<sub>3</sub>); mp 148–150 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) δ 1.79 (s, 9H), 2.02 (br s, 18H), 2.22 (s, 9H), 4.17 (d, J=5.4 Hz, 6H), 4.33 (t, J=6.3 Hz, 3H), 4.74 (br m, 6H), 5.30 (d, J=10.2 Hz, 3H), 5.56 (br s, 3H), 5.64 (t, J=9.75 Hz, 3H), 5.96 (d, J=9.3 Hz, 3H), 8.02 (s, 3H), 8.12 (br s, 3H), 8.21 (s, 3H); <sup>13</sup>C NMR  $(CDCI<sub>3</sub>, 75 MHz)$   $\delta$  20.2, 20.5, 20.67, 20.70, 35.5, 61.1, 66.8, 68.0, 70.8, 73.9, 86.2, 121.6, 128.7, 134.8, 145.4, 166.1, 169.3, 169.9, 170.1, 170.3; IR (KBr) 3390, 2929, 1750, 1660, 1534, 1430, 1373, 1224, 1056 cm<sup>-1</sup>; HRMS calcd for C<sub>60</sub>H<sub>72</sub>N<sub>12</sub>O<sub>30</sub>Na m/z 1463.4375, found 1463.4369.

4.28.1.2. Dendrimer **25b.** Yield: 90%;  $[\alpha]_D^{24.3}$  -31.53 (c 0.33, CHCl<sub>3</sub>); mp 132–134 °C; <sup>1</sup>H NMR (CD<sub>3</sub>OD, 300 MHz)  $\delta$  1.82 (s, 9H), 1.97 (s, 9H), 2.01 (s, 9H), 2.07 (s, 9H), 4.18 (br d,  $J=11.1$  Hz, 3H), 4.24–4.34 (br m, 6H), 4.74 (br m, 6H), 5.32 (t,  $J=9.1$  Hz, 3H), 5.56 (t,  $J=9.3$  Hz, 3H), 5.73 (t,  $J=9.1$  Hz, 3H), 6.19 (d,  $J=9.0$  Hz, 3H), 8.33 (s, 3H), 8.35 (s, 3H); IR (KBr) 3418, 2925, 2858, 1748, 1652, 1533, 1463, 1423, 1379, 1324, 1226 cm<sup>-1</sup>; HRMS calcd for C<sub>60</sub>H<sub>72</sub>N<sub>12</sub>O<sub>30</sub>Na m/z 1463.4375, found 1463.4375.

4.28.2. Synthesis of dendrimers  $27a$ , b. To an ice-cooled solution of hexaalkyne 26 (0.05 g, 0.064 mmol) in 20 ml of dry acetonitrile under nitrogen atmosphere was added diisopropylethylamine (0.076 ml, 0.446 mmol), azido sugar (24a or 24b) (0.167 g, 0.446 mmol), and CuI (0.01 g, 0.045 mmol). The reaction mixture was stirred under  $N_2$  atmosphere for 17 h and solvent was evaporated. The solid thus obtained was dissolved in ethyl acetate, washed with 0.2 N  $H_2SO_4$  water,  $NH_4Cl/NH_4OH$  (9:1) solution, and finally with water. The organic layer was dried over anhydrous Na2SO4, filtered, and evaporated to yield the dendrimer.

4.28.2.1. Dendrimer **27a**. Yield: 80%;  $[\alpha]_D^{26.1}$  –3.3 (c 0.18, CHCl<sub>3</sub>); mp 138–140 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.84 (s, 18H), 2.01 (m, 18H), 2.03 (m, 18H), 2.23 (s, 18H), 3.10 (m, 6H), 4.17 (m, 12H), 4.30  $(m, 6H)$ , 5.12  $(m, 3H)$ , 5.15-5.45 (br m, 18H), 5.55  $(m, 12H)$ , 5.92 (d, J=8.1 Hz, 6H), 7.97 (m, 9H), 8.41 (d, J=6.3 Hz, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz) d 20.2, 20.5, 20.7, 36.1, 49.6, 57.8, 58.5, 61.1, 61.2, 66.8, 67.9, 68.0, 70.7, 73.8, 73.9, 86.0, 86.1, 122.7, 123.1, 129.5, 134.4, 142.8, 165.7, 165.8, 169.2, 169.9, 170.1, 170.13, 170.2, 170.3, 170.4; IR (KBr) 3475, 2932, 1752, 1668, 1531, 1375, 1227, 1058 cm $^{-1}$ ; HRMS calcd for  $C_{123}H_{147}N_{21}O_{69}N$ a *m/z* 3044.8537, found 3044.8397.

4.28.2.2. Dendrimer 27b. Yield: 80%;  $\alpha$  $]_D^{26.6}$  – 29.2 (c 0.11, CHCl<sub>3</sub>); mp 140-142 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.75-1.92 (m, 18H),  $1.97 - 2.12$  (m, 54H), 3.08 (m, 6H), 4.00-4.22 (m, 12H), 4.24-4.38 (br m, 6H), 5.05-5.38 (br m, 21H), 5.45 (t, J=9.3 Hz, 6H), 5.51-5.62 (m, 6H), 5.96 (m, 6H), 7.90 (m, 3H), 7.90-8.15 (m, 6H), 8.37 (d, J=8.7 Hz, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz) δ 20.1, 20.5, 21.0, 36.0, 49.5, 57.8, 58.6, 60.4, 61.6, 67.7, 70.2, 72.6, 75.0, 85.6, 122.8, 123.4, 129.4, 134.4, 142.7, 142.9, 165.6, 168.9, 169.4, 170.0, 170.3, 170.5; IR (KBr) 3403, 2942, 1752, 1667, 1531, 1447, 1376, 1227, 1042  $\text{cm}^{-1}$ ; HRMS calcd for  $C_{123}H_{147}N_{21}O_{69}N$ a *m/z* 3044.8537, found 3044.8423.

## Acknowledgements

Financial assistance from CSIR and DST, New Delhi, is acknowledged. We thank Mr. Kashmiri Lal and Mr. K.Y. Kiran Kumar for their help in the synthesis of dendrimers.

## Supplementary data

<sup>1</sup>H NMR, <sup>13</sup>C NMR, and HRMS of all new compounds. Supplementary data related to this article can be found online at [doi:10.1016/j.tet.2011.01.023](http://dx.doi.org/doi:10.1016/j.tet.2011.01.023).

#### <span id="page-11-0"></span>References and notes

- 1. Mammen, M.; Cho, S.-K.; Whitesides, G. M. Angew. Chem., Int. Ed. 1998, 37, 2754-2794
- 2. Kano, K.; Liu, M.; Frechet, J. M. J. Bioconjugate Chem. 1999, 10, 1115-1121.
- 3. Grinstaff, M. W. Chem.-Eur. J. 2002, 8, 2838-2846.
- 4. Morosini, V.; Frochot, C.; Barberi-Heyob, M.; Schneider, R. Tetrahedron Lett. 2006, 47, 8745-8749.
- 5. Newkome, G. R.; Moorefield, C. N.; Baker, G. R. M.; Saunders, J.; Grossman, S. H. Angew. Chem., Int. Ed. Engl. 1991, 30, 1178-1180.
- 6. (a) Boas, U.; Heegaard, P. M. H. Chem. Soc. Rev. 2004, 33, 43-63; (b) Tomalia, D. A.; Naylor, A. M.; Goddard, W. A., III. Angew. Chem., Int. Ed. Engl. 1990, 29, 138-175; (c) Frechet, J. M. Science 1994, 263, 1710-1715.
- 7. Huisgen, R.; Koch, H. J. Liebigs Ann. Chem. 1955, 591, 200-231.
- 8. Rostovtsev, V. V. L.; Green, G.; Fokin, V. V.; Sharpless, K. B. Angew. Chem., Int. Ed. 2002, 41, 2596-2599.
- 9. For detailed reviews on this topic, (a) Bock, V. D.; Hiemstra, H.; van Maarseveen, J. H. Eur. J. Org. Chem. 2006, 51-68; (b) Meldal, M.; Torn $\oslash$ e, C. W. Chem. Rev. 2008, 108, 2952-3015.
- 10. (a) Horne, W. S.; Stout, C. D.; Ghadiri, M. R. J. Am. Chem. Soc. **2003**, 125, 9372–9376; (b) van Maarseveen, J. H.; Horne, W. S.; Ghadiri, M. R. Org. Lett.  $2005$ ,  $7,4503-4506$ .
- 11. Dawson, P. E.; Muir, T. W.; Clark-Lewis, I.; Kent, S. B. H. Science 1994, 266, 776-779.
- 12. (a) Saxon, E.; Bertozzi, C. R. Science 2000, 287, 2007-2010; (b) Nilsson, B. L.; Kiessling, L. L.; Raines, R. T. Org. Lett. 2000, 2, 1939-1941; (c) Saxon, E.; Armstrong, J. I.; Bertozzi, C. R. Org. Lett. 2000, 2, 2141-2143; (d) Nilsson, B. L.; Kiessling, L. L.; Raines, R. T. Org. Lett. 2001, 3, 9-12.
- 13. (a) Liu, C. F.; Tam, J. P. Proc. Natl. Acad. Sci. U.S.A. 1994, 91, 6584-6588; (b) Liu, C. F.; Tam, J. P. J. Am. Chem. Soc. 1994, 116, 4149-4153.
- 14. Muir, T. W.; Sondhi, D. P.; Cole, A. Proc. Natl. Acad. Sci. U.S.A. 1998, 95, 6705-6710
- 15. (a) Lewis, W. G.; Green, L. G.; Grynszpan, F.; Radic, Z.; Carlier, P. R.; Taylor, P.; Fin, M. G.; Sharpless, K. B. Angew. Chem., Int. Ed. 2002, 41, 1053-1056; (b) Chin, J. W.; Santoro, S. W.; Martin, A. B.; King, D. S.; Wang, L.; Schultz, P. G. J. Am. Chem. Soc. 2002, 124, 9026-9027.
- 16. (a) Wu, P.; Feldman, A. K.; Nugent, A. K.; Hawker, C. J.; Scheel, A.; Voit, B.; Pyun, J.; Frechet, J. M. J.; Sharpless, K. B.; Fokin, V. V. Angew. Chem., Int. Ed. 2004, 43, 3928-3932; (b) Opsteen, J. A.; Hest, J. C. M. V. Chem. Commun. 2005, 57-59.
- 17. (a) Twyman, L. J.; Beezer, A. E.; Mitchell, J. C. Tetrahedron Lett. 1994, 35, 4423-4424; (b) Sadler, K.; Tam, J. P. Rev. Mol. Biotechnol. 2002, 90, 195-229.<br>18. (a) Haridas, V.; Lal, K.; Sharma, Y. K. Tetrahedron Lett. 2007, 48, 4719-4722;
- (b) Haridas, V.; Sharma, Y. K.; Naik, S. Eur. J. Org. Chem. **2009**, 1570–1577. 19. (a) van Baal, I.; Malda, H.; Synowsky, S. A.; van Dongen, J. L. J.; Hackeng, T. M.;
- Merkx, M.; Meijer, E. W. Angew. Chem., Int. Ed. 2005, 44, 5052-5057; (b) Gutiérrez-Abad, R.; Illa, O.; Ortuño, R. M. Org. Lett. 2010, 12, 3148-3151.
- 20. (a) Martin, I. K.; Twyman, L. J. Tetrahedron Lett. **2001**, 42, 1119–1121; (b) Jiang, Z.-X.; Yu, Y. B. J. Org. Chem. **2010**, 75, 2044–2049.
- 21. (a) Heeres, A.; van der Pol, C.; Stuart, M.; Friggeri, A.; Feringa, B. L.; van Esch, J. J.Am. Chem. Soc. 2003, 125, 14252–14253; (b) van Bommel, K. J. C.; van der Pol, C.; Muizebelt, I.; Friggeri, A.; Heeres, A.; Meetsma, A.; Feringa, B. L.; van Esch, J. Angew. Chem., Int. Ed. 2004, 43, 1663-1667; (c) Love, C. S.; Hirst, A. R.; Chechik, V.; Smith, D. K.; Ashworth, I.; Brennan, C. Langmuir 2004, 20, 6580–6585; (d) de Loos, M.; van Esch, J. H.; Kellogg, R. M.; Feringa, B. L. Tetrahedron 2007, 63, 7285-7301; (e) Hirst, A. R.; Huang, B.; Castelletto, V.; Hamley, I. W.; Smith, D. K.  $Chem—Eur. I. 2007. 13, 2180-2188.$
- 22. (a) Dubber, M.; Lindhorst, T. K. Chem. Commun.  $1998$ ,  $1265-1266$ ; (b) Dubber, M.; Lindhorst, T. K. Org. Lett. 2001, 3, 4019-4022; (c) Ashton, P. R.; Balzani, V.; Clemente-Leon, M.; Colonna, B.; Credi, A.; Jayaraman, N.; Raymo, F. M.; Stoddart, J. F.; Venturi, M. Chem.-Eur. J. 2002, 8, 673-684; (d) Turnbull, W. B.; Kalovidouris, S. A.; Stoddart, J. F. Chem.-Eur. J. 2002, 8, 2988-3000; (e) Kiessling, L. L.; Gestwicki, J. E.; Strong, L. E. Angew. Chem., Int. Ed. 2006, 45, 2348-2368; (f) Kikkeri, R.; Liu, X.; Adibekian, A.; Tsai, Y.-H.; Seeberger, P. H. Chem. Commun. 2010, 2197-2199.
- 23. (a) Ashton, P. R.; Boyd, S. E.; Brown, C. L.; Jayaraman, N.; Stoddart, J. F. Angew. Chem., Int. Ed. Engl. 1997, 36, 732-735; (b) Zianini, D.; Roy, R. J. Am. Chem. Soc. 1997, 119, 2088-2095; (c) Vrasidas, I.; de Mol, N. J.; Liskamp, R. M. J.; Pieters, R. J. Eur. J. Org. Chem. 2001, 4685-4692.
- 24. (a) Shiozaki, M.; Mochizuki, T.; Hanzawa, H.; Haruyama, H. Carbohydr. Res. 1996, 288, 99-108; (b) Vicente, V.; Martin, J.; Jimenez-Barbero, J.; Chiara, J. L.; Vicent, C. Chem.-Eur. J. 2004, 10, 4240-4251.
- 25. Brik, A.; Alexandratos, J.; Lin, Y.-C.; Elder, J. H.; Olson, A. J.; Wlodawer, A.; Goodsell, D. S.; Wong, C.-H. ChemBioChem 2005, 6, 1167-1169.